

Date : 15/03/2021

NOTICE

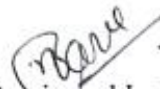
All the students of First Year Engineering Department are hereby informed that your First Sesimal Examination will be started from 19/03/2021 as per the given below time table.

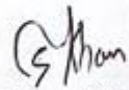
Instructions:-

- 1) Sessional Exam is taken on Google Form
- 2) Solve 15 MCQ's out of 20 MCQ's
- 3) Every question carries 2 marks
- 4) For any query contact your subject teacher
- 5) You have to submit the paper within given time up to 12.30 pm.


Section- A		
Subject	Date	Time
Computer Programming	19/03/2021	11:30 am to 12:30 pm
Engineering Physice	20/03/2021	11:30 am to 12:30 pm
Engineering Mechanics	21/03/2021	11:30 am to 12:30 pm
Mathematics-I	22/03/2021	11:30 am to 12:30 pm

Section- B		
Subject	Date	Time
Electrical Engineering	19/03/2021	11:30 am to 12:30 pm
Engineering Chemistry	20/03/2021	11:30 am to 12:30 pm
Engineering Drawing	21/03/2021	11:30 am to 12:30 pm
Mathematics-II	22/03/2021	11:30 am to 12:30 pm


Sessional Incharge
Dr. M. B. Wary


Prof. M. A. Khan
First Year Incharge




Principal
Jagadamba Collage of Engineering &
Technology Arni Road, Kinhi, Yavatmai

JAGADAMBHA COLLEGE OF ENGINEERING AND TECHNOLOGY, YAVATMAL
B.E.FIRST YEAR, FIRST SEMESTER SESSION 2020-21

First Sessional Examination Marks

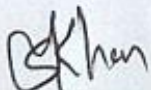
Sr.No.	Name of Student	Section	SESSIONAL FIRST				Total marks out of 120
			M-I	PHY	EM	CP	
1	AMISHA DAMESHWAR WETTI	A	28	26	30	30	114
2	YASHASHRI C GAYKI	A	30	30	30	30	120
3	SNEHA GOVIND GADHAVE	A	28	26	30	30	114
4	TEJAS RAMRAO PILLARE	A	30	26	30	30	116
5	SAKSHI SHAM SHARMA	A	24	26	26	30	106
6	RUTVIK RAJENDRA PAGRUT	A	28	30	28	22	108
7	SHRIPAD JAGDISH KHORGADE	A	30	30	30	30	120
8	SWAMINI MAHENDRA GAWANDE	A	30	30	30	30	120
9	RACHANA MADHAV INGOLE	A	30	30	26	30	116
10	SANIKA VIVEK PANDE	A	26	26	30	30	112
11	NIKITA GAJANAN INGLE	A	30	30	30	30	120
12	KHUSHI RAJENDRAKUMAR	A	30	30	30	30	120
13	ASIYA MOBINKHAN TIGALE	A	30	30	30	30	120
14	TRUPTI BHARAT BHAGAT	A	30	28	30	28	116
15	KALYANI ANIL TONCHAR	A	30	30	24	30	114
16	TRIVENI SUBHASH RATHOD	A	30	30	30	30	120
17	SAKSHITA ASHOK DHAKANE	A	24	28	30	30	112
18	ATUL SANTOSH RAHOD	A	30	30	30	30	120
19	MADHURA MANOJ PESHWE	A	30	30	30	30	120
20	CHINMAYI MUKUND TAMBULE	A	30	30	30	12	102
21	SAKSHI RAJENDRA KHANDARE	A	30	28	30	30	118
22	GAURAV NAMDEV GORLE	A	30	28	30	30	118
23	DARSHAN DILIP RATHOD	A	30	30	30	30	120
24	VISHAL SHANKARRAO	A	22	30	30	30	112
25	KETAN BABULAL NISWADE	A	28	30	30	30	118
26	NEHA SURESHRAO SARODE	A	30	30	22	30	112
27	SAMIKSHA MADHUKAR GOPAL	A	30	30	30	30	120
28	SANDHYANI KAILAS PATE	A	30	30	24	20	104
29	DIKSHA ARVIND TAYADE	A	28	30	22	30	110
30	VRUSHABH GOPAL HIWARALE	A	30	28	30	30	118
31	MAHESH HANUMAN JAISWAL	A	24	26	30	30	110



32	ROHIT BABULAL MUDRAKE	A	26	24	22	30	102
33	BHAVIKA UKANDRAO TODSAM	A	30	24	28	30	112
34	PRATIK G KHANDWE	A	30	28	30	30	118
35	SAJAN UTTAM ADE	A	24	28	28	30	110
36	PRAJWAL VILAS DUDHE	A	30	26	26	30	112
37	PRASHANT DINESH KAPATE	A	22	30	0	30	82
38	PRANAV GAJANAN PARLAWAR	A	30	30	30	30	120
39	ISHA G. KHARATE	A	30	30	30	22	112
40	VASUNDHARA PRAKASH	A	30	26	28	30	114
41	SAKSHI NANDKISHOR THORAT	A	30	28	30	30	118
42	MANISHA PRABHATILAL	A	26	30	28	30	114
43	UMESH RAJU SHINDE	A	30	30	26	30	116
44	RUTIK SANJAY INGOLE	A	30	30	30	30	120
45	SANIKA RAJESH KOLWADKAR	A	30	30	30	30	120
46	RANI KISANRAO NANDANE	A	22	28	30	30	110
47	RENUKA DINESH BADE	A	26	24	28	30	108
48	MANAV INDRAKANT GHUGARE	A	30	30	20	30	110
49	SUJAL RAMBHAU GHAYAL	A	22	30	30	24	106
50	GAYATRI DIPAK GADEKAR	A	30	30	30	30	120
51	SHIVAM AMOL TURKHADE	A	24	30	0	30	84
52	BHAVNA VIJAY DAHAKE	A	30	30	20	30	110
53	AMRUTA VINOD THAVRI	A	26	30	24	8	88
54	SANI SURESH RATHOD	A	28	30	30	30	118
55	DIPAK RAMESH GAWAI	A	28	24	12	30	94
56	GHATOLE AKSHAY DILIP	A	26	20	26	30	102
57	SAKSHI KAKADE	A	13	14	15	15	57
58	MAYURI JAISINGH JADHAO	A	30	6	30	7	73
59	SAKSHI V. RAUT	A	13	8	7	14	42
60	PRITI DILIP RATHOD	A	13	13	14	20	60
61	SANGHARSH LOHAKARE	A	24	0	0	30	54
62	SANJANA KISAN CHAKRE	A	14	12	13	15	54
63	RUTUJA SATISH MANCHALWAR	A	5	28	6	7	46
64	KARAN SURESH CHAVHAN	A	25	0	0	24	49
65	TANMAY SHANKARRAO CHAVHAN	A	3	20	14	16	53


Sessional Incharge




First Year Incharge

Sr.No.	Name of Student	Section	M-II	Chem	EG	EE	Total marks out of 120
1	NIKHIL UDEBHANJI NAGOSE	B	30	30	30	30	120
2	Gaurav P. Shrivastav	B	30	30	30	30	120
3	SAKSHI DILIP BHOKTE	B	30	30	30	30	120
4	ABHAY RAJKUMAR MAHALLE	B	30	26	30	30	116
5	ALTAF KHUDDUS BAIG	B	30	22	30	30	112
6	PRATHAMESH KINHIKAR	B	30	30	30	30	120
7	ROHIT ASHOK PAWAR	B	30	24	30	30	114
8	SHREYA ARUN THAWARI	B	30	28	30	30	118
9	SWEETY RAJU BHOKTE	B	22	30	30	30	112
10	Abhishek Kishor Bhojar	B	30	26	30	30	116
11	RAKSHANDA PANDIT JADHAO	B	30	24	30	30	114
12	RUTIK BAGMARE	B	28	24	30	30	112
13	JAY SUNIL BHOYAR	B	30	28	30	30	118
14	Rushant Dinkar Bobde	B	24	28	30	30	112
15	Rohan Mahadeo Chavhan	B	28	26	22	22	98
16	RAJ SUDHIR PAWAR	B	26	30	30	30	116
17	EKTA SUDESH RATHOD	B	30	28	30	30	118
18	SAHIL MANUTRAO INGULKAR	B	30	26	30	30	116
19	VAISHNAVI SANTOSH ZAGALE	B	30	24	30	30	114
20	TEJAS SANTOSH BOHARE	B	30	28	30	30	118
21	SHIVANI WANKHADE	B	26	22	30	30	108
22	ABHISHEK KRUSHNA MUNGLE	B	24	24	30	30	108
23	SHEKH AYYAS SHEKH AHEMAD	B	30	24	30	30	114
24	KUNAL JITENDRA GIRI	B	28	28	30	30	116
25	SHREYA WANKHADE	B	30	24	30	30	114
26	PRATHMESH SARODE	B	22	30	30	30	112
27	GOPAL VINOD GAWANDE	B	28	22	30	30	110
28	TANU TULSIDAS PATEL	B	26	30	30	30	116
29	TWINKLE SIDDHARTH LADHE	B	30	30	30	30	120
30	DARSHAN SURESH BURADE	B	30	30	30	30	120
31	TEJAS RAVINDRA KEVATE	B	30	30	30	30	120
32	HARSH RAMESH PALI	B	30	30	30	30	120
33	SNEHAL VINAYAK MORE	B	30	20	30	30	110
34	Chetan R. Sonvane	B	24	24	30	30	108
35	SANIYA ARNAB SYED ALI	B	30	26	30	30	116
36	ARPITA VINAYAK BHAVARE	B	28	26	30	30	114
37	ROSHAN DEVIDAS RATHOD	B	24	22	30	30	106
38	HIMANSHU ANIL JADHAO	B	26	24	30	30	110
39	SUKANYA BHIMRAO PATIL	B	30	28	30	30	118
40	DANISH MAHEMOOD SHEIKH	B	28	14	30	30	102
41	YASH PRABHAKAR DHOTE	B	30	24	30	30	114



42	SHIVLING NAGORAO DUPARTE	B	22	26	30	30	108
43	NIRAJ UTTAM PAWAR	B	30	24	30	30	114
44	DNYANESHWAR JADHAV	B	30	18	30	30	108
45	POOJA HARI NANDANE	B	28	24	30	30	112
46	SHUBHAM BADGHAIYA	B	30	24	30	30	114
47	MAHESH SUBHASH RATHOD	B	24	26	30	30	110
48	SONALI SANTOSH CHAVHAN	B	30	28	30	30	118
49	DEEP RAVINDRA TAJNE	B	30	26	30	30	116
50	DIGVIJAY MEHATRE	B	30	18	28	28	104
51	Dhyaneshwar Rathod	B	30	26	30	30	116
52	Ganesh Kanade	B	30	24	30	30	114
53	ADARSH RAJESH DAMLE	B	30	24	30	30	114
54	NIKITA TUKARAM RATHOD	B	30	26	30	30	116
55	ASIT DNYANESHWAR JAMNIK	B	26	24	30	30	110
56	VAISHNAVI TUMBADE	B	30	22	30	30	112
57	RAHUL PRAKASH HAWALDAR	B	30	28	30	30	118
58	PALLAVI UKANDA JADHAO	B	30	28	12	12	82
59	BHAGYASHREE RATHOD	B	26	20	8	30	84
60	SACHIN MAROTI ADE	B	24	24	30	30	108
61	EKTA DASHRATH CHAVHAN	B	30	26	30	30	116
62	SAKSHI ASHOK KAMBLE	B	30	28	30	30	118
63	AKSHAY FULMALI	B	30	22	20	20	92
64	SANDESH MOHAN RATHOD	B	22	18	24	24	88
65	SANTOSH GAJANAN KANDALE	B	13	13	15	12	53
66	ROSHANI DURGASING RATHOD	B	13	12	12	12	49
67	SAQIB SHAH BASHIR SHAH	B	20	14	3	3	40
68	Aditya V. Bhagat	B	14	12	14	15	55
69	VIKAS SANTOSH RATHOD	B	12	12	13	13	50

M. B. Wani
 Sessional Incharge
 Dr. M. B. Wani

M. A. Khan
 First Year Incharge M. A. Khan



H. P.
 Principal
 Jagadamba College of Engineering &
 Technology, Kinhi, Yavatmal

Jagdambha College of Engineering and Technology Yavatmal

Department= First Year Session 2020-21 Sem I Subject - Engineering Chemistry
Test

* Required

1. Email *

2. Name of Student *

3. Section *

4. Roll No. *

5. Contact Number (Student) *

6. Contact Number (Parent) *



7. 01. Hardness of water is present due to *

2 points

Mark only one oval.

- Salts
- solution
- gases
- air

8. 02. Alkaline hardness caused by *

2 points

Mark only one oval.

- bicarbonates of sodium
- carbonates of pottasum
- bicarbonates of Ca and Mg
- bicarbonates of Cu

9. 3. Units of hardness of water is *

2 points

Mark only one oval.

- gram/liter
- parts per million
- pounds
- grains

10. 4. Hardness of ware can be determined by *

2 points

Mark only one oval.

- Zeolite method
- Lime-Soda method
- EDTA method
- Ion-exchange method



11. 5. Zeolite softener possess *

2 points

Mark only one oval.

- exchangeable K ions
- exchangeable Na ions
- exchangeable H ions
- exchangeable Cr ions

12. 6. Permanent hardness is due to presence of *

2 points

Mark only one oval.

- bicarbonates of Ca/Mg ions
- oxides of nitrogen
- chloride/sulphate/nitrate salts of Ca/Mg ions
- carbonates of Fe

13. 7. Exhausted Zeolite softner can be regenerated by *

2 points

Mark only one oval.

- HCl solution
- NaOH solution
- NaCl solution
- KOH solution

14. 8. when hard water is passed through Cation exchange resin *

2 points

Mark only one oval.

- only acidic nature of water get removed
- only temporary hardness is removed
- temporary and permanent hardness get removed
- turbidity get removed



15. 9. Deionised water obtained from Ion-exchange process contain residual hardness up to * 2 points

Mark only one oval.

- 10 ppm
- 20 ppm
- 50 ppm
- 0-2 ppm

16. 10. RO water can be obtained by * 2 points

Mark only one oval.

- Reverse Osmosis process
- Ion exchange process
- Osmosis process
- Zeolite process

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A handwritten signature in blue ink, appearing to be "M.P.", written over a horizontal line.

Principal
Jagadambha Collage of Engineering &
Technology Arni Road, Kinhi, Yavatmal